

# Chapter 11

## Wine Spoilage by Fungal Metabolites

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### 11.1 Introduction

This chapter is devoted to the description of toxic and spoiling secondary metabolites produced by moulds and yeasts in grapes and wines. The toxic compounds affect human health being a food safety issue while spoilage concerns wine organoleptical quality being, therefore, a technological matter. Amongst several compounds with detrimental effects on human health reported as occurring in wine – lead, pesticides, ethyl carbamate, biogenic amines, and more recently ochratoxin A (OTA) – only the last is of fungal origin and the most preoccupying of them. The occurrence

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01 of OTA in grapes and wines has a special meaning in enology, because wine has  
02 been known, since remote antiquity, as “the safest beverage”. In fact, the enormous  
03 success of wine (and beer) for millennia in the Mediterranean basin is due to the  
04 bad sanitary quality of water, frequently contaminated by salmonella or other faecal  
05 pathogens around urban agglomerates. In the second half of the nineteenth century,  
06 Louis Pasteur stated that “Wine is the most hygienic of the beverages”, implying that  
07 any pathogenic bacteria had capacity to grow or survive in wine. The occurrence of  
08 OTA in wine changes the paradigm of “the safest beverage”, thus imposing a food  
09 safety concern to the wine industry.

10 Microorganisms have long been known as one of the main sources of wine  
11 spoilage. Earlier enology treatises mention the spoiling activities of lactic and acetic  
12 acid bacteria and yeasts, concerns being preferentially directed to the deleterious  
13 effects of acetic and lactic acid bacteria (Ribéreau-Gayon et al. 2006). More  
14 recently, yeasts have become a source of serious trouble for winemakers, especially  
15 after the confirmation of *Dekkera/Brettanomyces bruxellensis* as the producers of  
16 unwanted levels of 4-ethylphenol (4-EP) leading to phenolic or “horse-sweat” taints.  
17 These yeast species and lactic acid bacteria are also responsible for the production  
18 of tetrahydropyridines (THP), connected with the taint described as “mousiness”.

19 This chapter outlines the most recent advances in the awareness of the problems  
20 raised by ochratoxin A, volatile phenols and tetrahydropyridines. We will focus on  
21 the factors leading to their occurrence, the relevance to wine quality and safety and  
22 the respective preventive measures suitable under vineyard and winery practices.

## 24 11.2 Ochratoxin A

26 Mycotoxins are metabolites produced by fungi and affecting human health, causing  
27 mycotoxicoses. These diseases have been known for a long time but only after the  
28 early 1960s have they been appropriately studied (Bennett and Klich 2003; van  
29 Egmond and Schothorst 2007). The first mycotoxin to be studied was aflatoxin.  
30 Since then the number of known mycotoxins have increased to 300 or 400 com-  
31 pounds of variable toxicity and occurrence. The main food commodities affected  
32 are cereals used for direct human consumption, for processing, or as animal feeds  
33 and dried fruits (van Egmond and Schothorst 2007). Some mycotoxins have been  
34 mentioned in grape products (patulin, aflatoxins, trichothecenes) but OTA is the  
35 main toxin of concern in the wine industry (Hocking et al. 2007). The occurrence of  
36 OTA in wine was reported for the first time in 1995 (Zimmerli and Dick 1996) and  
37 much data has been generated since then for wine, grape juice and raisins, especially  
38 in the last seven years.

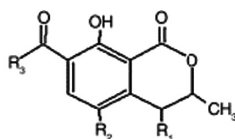
### 42 11.2.1 Chemical Structure

44 Ochratoxin A was originally described as a metabolite of the mould *Aspergillus*  
45 *ochraceus* in a laboratory screening for toxinogenic fungi (van der Merwe et al. 1965).

It is the most abundant and hence the most commonly detected member of the family of ochratoxins produced as secondary metabolites by moulds. Apart from ochratoxin  $\alpha$  ( $OT\alpha$ ), the ochratoxins comprise a polyketide-derived dihydroisocoumarin moiety linked via the 7-carboxy group to l- $\beta$ -phenylalanine by an amide bond. Ochratoxins consist of ochratoxin A (OTA), its methyl ester and its ethyl ester, also known as ochratoxin C (OTC), 4-hydroxyochratoxin A (4-OH OTA), ochratoxin B (OTB) and its methyl and ethyl esters and ochratoxin  $\alpha$  ( $OT\alpha$ ), where the phenylalanine moiety is missing (Fig. 11.1). All of them behave like weak organic acids and the differences in the chemical structures have marked effects on their respective toxic potentials, OTA being the most toxic of the group (Ringot et al. 2006). The presence of chlorine in the OTA structure makes it unique among mycotoxins (Murphy et al. 2006). The empirical formula is  $C_{20}H_{18}O_6NCl$  and the molecular weight is 403.82. The IUPAC developed formula of OTA is l-phenylalanine-*N*-[(5-chloro-3,4-dihydro-8-hydroxy-3-methyl-1-oxo-1*H*-2-benzopyran-7-yl)carbonyl]-(*R*)-isocoumarin (Ringot et al. 2006). The present knowledge of the biosynthetic pathway of OTA, which has not yet been fully established, has been reviewed by Ringot et al. (2006).

### 11.2.2 Toxicity and Public Health Safety

OTA is a mycotoxin considered to be a possible carcinogen (Class 2B) for humans (IARC 1993) and it has been shown to be nephrotoxic, hepatotoxic, teratogenic, carcinogenic and immunotoxic to several species of animals and to cause kidney and liver tumours in mice and rats (JECFA 2001). In humans it is accumulated in body tissues because it appears to be slowly eliminated, but the effects are not well established (Ringot et al. 2006). Ruminant animals, such as cows and ewes, are generally resistant to the effects of OTA due to its hydrolysis to non-toxic metabolites by protozoa in the stomachs before absorption into the blood (Kießling et al. 1984). Recent studies on the toxic mechanisms are focused on the OTA ability to disturb cellular signalling and regulation and to modulate physiological signals and thereby



Ochratoxins	R1	R2	R3
OTA	H	Cl	-NH-CH(COOH)-CH <sub>2</sub> -Phenyl
OTB	H	H	-NH-CH(COOH)-CH <sub>2</sub> -Phenyl
OTC	H	Cl	-NH-CH(COOC <sub>2</sub> H <sub>5</sub> )-CH <sub>2</sub> -Phenyl
4-hydroxyochratoxin A	OH	Cl	-NH-CH(COOH)-C <sub>2</sub> H-Phenyl
$OT\alpha$	H	Cl	-OH

Fig. 11.1 Chemical structure of ochratoxins (reprinted from Ringot et al. 2006, permission to be obtained)

01 to influence cells viability and proliferation, but its modes of toxicity remain “a  
02 continuing enigma” (see review of Ringot et al. 2006).

03 Given their toxicity, mycotoxins are subjected to regulations determining the  
04 maximum allowable levels. As a rule food products have mycotoxin levels lower  
05 than the limits (Jorgensen 2005) but about 40% of the notifications received in 2005  
06 by an European rapid alert system for food and feed were related to risks to human  
07 health by mycotoxins (van Egmond and Schothorst 2007). These authors further  
08 mentioned that almost 90% of these notifications were related to aflatoxin in nuts  
09 and nut products imported to the EU. Cases related to OTA in wine were not referred  
10 to in this report. In addition, an indirect measure of the toxicity and risk to human  
11 health may be given by the standards of international legislation or guidelines of  
12 advisory boards. Concerning OTA there are no regulations in the USA or in Codex  
13 Alimentarius Commission, in contrast to patulin, fumonisin, aflatoxin and DON  
14 (Murphy et al. 2006), and so it seems admissible that these toxins represent a bigger  
15 threat than OTA.

16 At European Union level the Council Regulation (EEC) 315/93 of 8th February  
17 1993 provided the legal framework for establishing maximum levels for food con-  
18 taminants at Community level. In 1995, the European Commission (EC) initiated  
19 the activity SCOOP (scientific cooperation on questions related to food), which  
20 included a project to provide data on the occurrence of OTA in food commodities  
21 on the European market and on the dietary exposure to OTA in the EU member  
22 states (Jorgensen 2005). As a consequence, many data on the occurrence of OTA  
23 in human food and human blood plasma have become available since 1995. After  
24 the first SCOOP report, known as SCOOP-1 (European Commission 1997) a second  
25 SCOOP task was performed, extended to other commodities and processed  
26 foods, including wine and other grape products, to evaluate if the additional studies  
27 changed the conclusions of first SCOOP report. Not surprisingly, due to the detec-  
28 tion of OTA in wine, much occurrence data has been produced since then, not only  
29 for wine but also for dried vine fruits (currants, raisins, and sultanas) and grape  
30 juice, particularly after 2000. The most relevant conclusion from those data was  
31 that the overall mean level of OTA in wine was 0.36  $\mu\text{g}/\text{kg}$  (mean of 1470 samples),  
32 representing the second source, after cereals, to the OTA exposure in the European  
33 diet (European Commission 2002), raising relevant concern and electing OTA as a  
34 threat to the European wine industry. In particular, JECFA (2001) calculated that  
35 the human OTA exposure was of 58%, 21%, 7%, 5% and 3% of total OTA intake  
36 for cereals, wine, grape juice, coffee and pork, respectively (Murphy et al. 2006).  
37 However, the calculation of these figures was based on the controversial assumption  
38 that the mean intake is represented by the arithmetic mean value. If the median  
39 value was used, the value of OTA in red wine would have been only 0.02  $\mu\text{g}/\text{kg}$   
40 and the contribution of wine consumption for OTA intake-rate would drop to 2%  
41 (Otteneder and Majerus 2000). The maximum allowable limits for OTA in several  
42 food products have been established recently in the EU, being the adult’s strictest  
43 value applied to wine (Table 11.1). In addition, a provisional tolerable weekly intake  
44 (PTWI) of 100–120 ng/kg body weight (bw) is advised (JECFA 2001) that was set  
45 based on a safety factor of 450 related with the renal function deterioration of pigs,

**Table 11.1** European Union and Swiss regulations for ochratoxin A ( $\mu\text{g}/\text{kg}$ )

Product	Maximum allowable concentration	Reference
Unprocessed cereals	5	Commission Regulation (2006)
All products derived from unprocessed cereals, including processed cereal products and cereals intended for direct human consumption with the exception of foodstuffs	3	
Dried vine fruit (currants, raisins and sultanas)	10	
Roasted coffee beans and ground roasted coffee, excluding soluble coffee	5	
Soluble coffee (instant coffee)	10	
Wine (including sparkling wine, excluding liqueur wine and wine with an alcoholic strength of not less than 15 vol.%) and fruit wine	2	
Aromatised wine, aromatised wine-based drinks and aromatised wine-product cocktails	2	
Grape juice, concentrated grape juice as reconstituted, grape nectar, grape must and concentrated grape must as reconstituted, intended for direct human consumption	2	
Processed cereal-based foods and baby foods for infants and young children	0.5	
Dietary foods for special medical purposes intended specifically for infants	0.5	
Coffee products (Switzerland)	8	Taniwaki (2006)

for which the lowest observed effect level (LOEL) was 0.008 mg/kg bw per day (Bakker and Pieters 2002). Current levels of weekly exposure to OTA in adults of EU member states vary between 15 and 60 ng/kg bw which is, at most, half quantity of the PTWY (CONTAM 2006).

Further evaluation of OTA occurrence in wines worldwide has revealed that variable proportions of wines are contaminated but only a rather small number has levels over the maximum allowable limit of 2.0  $\mu\text{g}/\text{kg}$  (Ng et al. 2004; Mateo et al. 2007). Higher concentrations are more frequent in wines produced from dried grapes and in raisins, but are less consumed than table wines (Burdaspal and Legarda 2007; Valero et al. 2007b). Burdaspal and Legarda (2007) showed that sweet wines only contributed with 3.1–3.8% (regular consumers) or 0.3–0.4% (whole adult population) to the PTWI. Therefore, wines do not seem to have a significant contribution to human exposure to OTA.

### 11.2.3 Occurrence

Ochratoxin A is among the most important mycotoxins affecting foods being described in food commodities together with aflatoxin, deoxyvalenol, T-2 toxins

01 and trichotecenes, although the relative weight of each other in public health has  
02 not been mentioned (see reviews of Frisvad et al. 2006 and Richard 2007). As in  
03 other mycotoxins, OTA can contaminate a wide variety of food commodities as a  
04 result of fungal infection in crops, in the field during growth, at harvest, in storage,  
05 and in shipment under favourable environmental conditions, especially when the  
06 foods are not properly dried. Then, if used in animal's feed it may contaminate their  
07 products (Murphy et al. 2006). OTA occurs mainly in the storage of cereals and  
08 grains under conditions that favour mould growth and toxin production. Therefore it  
09 is associated with cereal and cereal products, soy products, coffee and cocoa prod-  
10 ucts (CAST 2003; Jorgensen 2005; Frisvad et al. 2006). Meat and meat products  
11 such as salami and hams, hard cheeses, spices and beer have also been related with  
12 OTA (Jorgensen 2005; Frisvad et al. 2006). Dried fruits (dates, plums, apricots, figs,  
13 sultanas) are typical affected commodities (Iamanaka et al. 2006). OTA may also  
14 occur in house dust and other airborne particulates (Richard 2007).

15 Concerning the wine industry, OTA has been detected in wines, grape juice,  
16 vinegar and raisins (Jorgensen 2005; Varga and Kozakiewicz 2006). These products  
17 were the last to be associated with OTA but studies in wines have greatly increased  
18 since the first report in 1995, being the OTA related food commodity most studied  
19 in the last decade (Jorgensen 2005), demonstrating the effort put into the solution  
20 of the problem by the wine sector. Table 11.2 lists the results of recent relatively  
21 large surveys worldwide demonstrating, as already mentioned, that wines with lev-  
22 els higher than the maximum allowable concentration of 2.0 µg/L are not frequent.  
23 As a rule, the OTA concentration increases from white and rosé to red wines, from  
24 cold to warmer regions, and dessert or sweet wines have higher mean levels (Varga  
25 and Kozakiewicz 2006 and Table 11.2). We are not aware of large surveys of OTA  
26 levels in grapes but raisins and sultanas have higher mean OTA levels than fresh  
27 grapes (Varga and Kozakiewicz 2006; Iamanaka et al. 2006).

## 30 **11.2.4 Ochratoxin A Production on Grapes**

### 32 **11.2.4.1 OTA Producing Species**

34 Ochratoxin A of food commodities is produced by a small number of fungal species  
35 in the genera *Penicillium*, *Aspergillus* and the *Aspergillus* teleomorphs *Petromyces*  
36 and *Neopetromyces* (Frisvad and Samson 2000; Frisvad et al. 2004).

37 The main species that occur in grapes and, consequently, in grape juices, raisins,  
38 wine and wine derivatives belong to the so-called black aspergilli, taxonomically  
39 included in the *Aspergillus* section *Nigri*. Unfortunately, the taxonomy of this sec-  
40 tion is not completely known, creating many difficulties on the identification of  
41 strains, originating a proliferation of taxa, including species, subspecies, and vari-  
42 eties (for a discussion see Samson et al. 2004 and Frisvad et al. 2006). Based on phe-  
43 notypic comparisons of a broad collection of black aspergilli, Samson et al. (2004)  
44 considered 15 species provisionally accepted in *Aspergillus* section *Nigri*, four  
45 of those producing OTA and only two occurring on grapes, raisins and in

**Table 11.2** Occurrence and concentration of OTA ( $\mu\text{g/L}$ ) in wines obtained in recent large scale surveys

Wine type	Number of samples	Percentage of positive samples	Percentage of samples > $\mu\text{g/L}$	Range	Mean	Median	Reference
White	60	25	0	<0.010–1.36	0.108	0.01	Otteneder and Majerus (2000)
Rosé	55	40	–	<0.010–2.38	0.119	0.01	
Red	305	54	–	<0.010–3.31	0.201	0.02	
Red (North Italy)	17	70	0	<0.010–0.54	0.12	0.05	Perrone et al. (2007)
Red (Central Italy)	46	59	0	<0.010–0.80	0.07	0.02	
Red (South Italy)	49	100	18	0.02–4.93	1.36	1.03	
Sweeta	290	96.9	6.8	<0.010–4.63	0.499	0.14	Burdaspal and Legarda (2007)
Sweet (A, B, CI) <sup>b</sup>	20	0	0	<0.024	<0.024	–	Valero et al. (2007b)
Sweet (CII)	26	54	15.3	<0.024–27.79	2.01	–	
Sweet (CIII)	75	60	26.7	<0.024–15.62	1.71	–	

<sup>a</sup>Maximum allowable levels for sweet and fortified wines are not legislated (see Table 11.1)

<sup>b</sup>European regions of wine production based on production conditions, soil, region and climate (Valero et al. 2007b)

wine – *A. carbonarius* and to a lesser extent *A. niger*. Recently, a new species of section *Nigri* isolated from grapes and OTA non-producer – *A. ibericus* – was proposed (Serra et al. 2006), being likely that new species can enlarge that section in the near future. In fact, with recent work on the molecular characterization of a Southern Europe population of black aspergilli isolated from grapes it was concluded that these represent a complex of species, where some of them are peculiar to grapes (Perrone et al. 2006b). Another study performed with strains isolated from grapes in Italy indicated that *A. tubingensis* is able to produce OTA and that, together with *A. carbonarius* and *A. niger*, it may be responsible for the OTA contamination of Italian wines (Perrone et al. 2006a). Some references in the literature also describe the occurrence of the yellowish *A. ochraceus* (belonging to the *Aspergillus* section *Circumdati*) and the blue-green *Penicillium* species on grapes (Frisvad et al. 2004), but its importance seems to be minor when compared with black aspergilli.

The evaluation of OTA production by fungi on infected grapes is essential to establish the real producing ability by the different species because the results of tests on synthetic culture media are not always coincident with in vivo determinations (Bellí et al. 2007). Overall, most *A. carbonarius* strains have the ability to produce OTA in grapes whereas the proportion is lower in the other toxigenic species (Perrone et al. 2006a, 2006b).

From a scientific point of view, precise strain identification of OTA fungi producers is important to establish the phylogenetic relationships among species, to recognize the mycota of foodstuffs, and to understand the peculiarities and ecological needs of the species. However, from a viticultural and enological point of view, it will be much more important to establish the environmental conditions that are ideal for OTA fungi producers on grapes than their identification.

#### 11.2.4.2 Factors Affecting Fungal Growth and OTA Production

The primary sources of *A. carbonarius* and *A. niger* are soil, bunch remnants or vine trash on soil vineyards, which are transported by wind from soil onto berry surfaces (Leong et al. 2006a; Hocking et al. 2007). Generally, the colonisation of grape bunches by black aspergilli and other fungi occurs when berry skin damage allows the entry into fruit tissues, where the low pH and high sugar content under aerobic conditions provide a competitive advantage for moulds. However, fungal invasion may occur without visible symptoms (Bellí et al. 2007). As a rule, the competition among contaminant microorganisms is more favourable to *Botrytis cinerea*, the agent responsible for common grey rot. However, other rot processes, such as sour rot and brown rot, can occur together with black aspergilli. The population dynamics of fungi outside or inside grape berries with skin damage is still poorly known, as it is difficult to establish which environmental conditions promote the dominance of each fungal species. Nevertheless, it has been shown that high temperatures (30 °C) and high relative humidity, between 80% and 100%, give rise to higher amounts of OTA produced by *A. carbonarius* on grapes (Bellí et al. 2007), suggesting that such conditions give competitive advantages to the black aspergilli population. Medina (2007) also showed that *A. carbonarius* growth was favoured



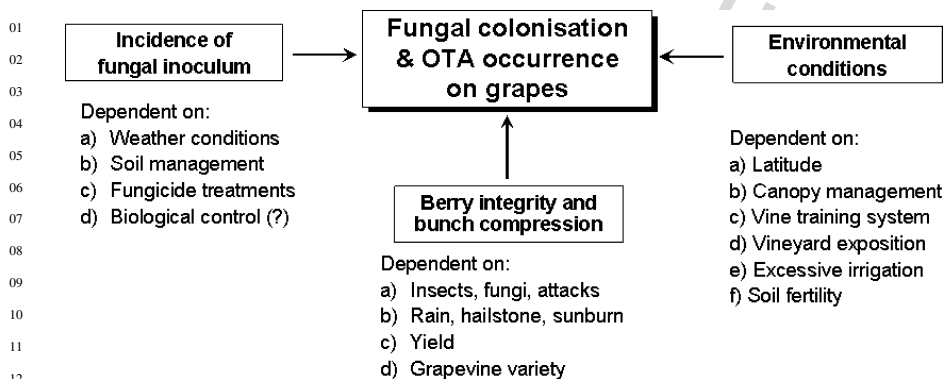
01 by high  $a_w$  (0.98) and temperature (28 °C), whereas OTA production was increased  
02 at mild temperature (20 °C) and 0.96–0.98  $a_w$ . In addition, Astoreca et al. (2007)  
03 found that optimum conditions of  $a_w$  and temperature are more restrictive for OTA  
04 production than for *Aspergillus* growth. The influence of relative humidity, rain or  
05 high  $a_w$  is less important than the influence of high temperature (Bellí et al. 2005;  
06 Medina 2007), explaining the occurrence of these fungi in hot and dry climate  
07 regions (Hocking et al. 2007). The influence of  $a_w$  appears to be more important  
08 in dried grapes, where black sultanas ( $a_w = 0.629$ ) showed levels of OTA higher than  
09 10  $\mu\text{g}/\text{kg}$  when white sultanas ( $a_w = 0.567$ ) did not (Iamanaka et al. 2006).

10 The frequency of the occurrence of OTA producing species on grapes is then  
11 essentially limited to conditions of high humidity and temperature, typical of sub-  
12 tropical and Mediterranean climates. In fact, OTA has been detected on grapes  
13 produced in France, South America, Spain, Italy, Portugal, Greece and Australia  
14 (as revised by Leong et al. 2006a), that have some wine regions with climatic  
15 conditions favouring black aspergilli species. In general, *A. carbonarius* is highly  
16 dominant, particularly in warmer regions, because their black spores are resistant  
17 to UV light and sun-drying (Leong et al. 2006a). However, in colder regions such  
18 as Germany, Northern Hungary, the Czech Republic or Northern parts of Portugal,  
19 France and Italy, black aspergilli have not been isolated from grape berries in spite  
20 of the presence of OTA in wines (for references see Blesa et al. 2006 and Varga  
21 and Kozakiewicz 2006) suggesting that other species, mainly *Penicillium*, adapted  
22 to cool temperatures, should be involved.

23 Zimmerli and Dick (1996) were the first authors to show that OTA content in  
24 southern wine-growing regions are higher than those of wines from northern areas,  
25 results also supported by the extensive work of Otteneder and Majerus (2000). More  
26 recently, OTA has been detected on grapes produced in many wine countries, where  
27 the highest amounts of OTA detected in each survey is generally correlated to vines  
28 growing in the warmest regions of each country (Blesa et al. 2006 and Table 11.2).  
29 However, the above-mentioned assumptions were not observed in Australia and  
30 South Africa, where no correlation was found between OTA incidence and wine  
31 region climate (Leong et al. 2006a). This fact was probably due to the rather low  
32 detected levels making these correlations imprecise. Further results in Australia  
33 showed apparently lower incidences in the cooler climate of Tasmania (Hocking  
34 et al. 2007). Canadian wines and grape juices have OTA levels comparable with data  
35 from cold climate wine growing areas in Europe (Ng et al. 2004 and Table 11.2).  
36 The fact that sweet wines from colder regions do not show high levels of OTA in  
37 contrast to those from warmer climates (Valero et al. 2007b) indicates that the main  
38 factor determining the OTA concentration in wines is the contamination of grapes  
39 by toxinogenic moulds.

### 41 42 **11.2.5 Prevention and OTA Production Control on Grapes**

44 Given the ubiquity of black aspergilli in vineyards of warm regions, all agents  
45 involved in the wine industry must learn how to live with them, minimising the



**Fig. 11.2** Factors affecting grape colonisation by OTA producer fungi

effect of all factors – biotic and abiotic – that contribute to their infection and growth on grape berries (see Fig. 11.2). Tillage operations contribute to the spreading of fungal spores and should be minimised when possible (Leong et al. 2006a). Concerning fungicide treatments, although some compounds have shown its efficiency, others seem to stimulate OTA production (Magan 2006) and so the most significant function of agrichemicals is to avoid grape damage by phytopathogenic moulds or insects which provide an easy entry for spore infection. Recently, a Code of Good Viticultural Practices was recommended by the Office International de la Vigne et du Vin (OIV 2005) to be put in use mainly in critical areas of *A. carbonarius* occurrence. Prevention is especially important for dried grapes, either for direct consumption or for dessert wines, given their higher levels of OTA. The main measures to consider were the following, mainly directed to avoid berry damage:

- To avoid all the cropping practices that lead to an excessive vigour of vine plants and an exaggerated increase in the yield, which make bunches more compact and thereby susceptible to berry splitting (Mínguez et al. 2004; Leong et al. 2006a)
- To avoid the use of cultivars with thin skinned berries and too compact bunches (Bellí et al. 2007)
- To carry out a plant sanitary program with efficient agrichemicals against moulds, particularly powdery mildew and grey rot, and insects, mainly *Lobesia* spp., using the adequate dose and timing, and making sure that the active ingredients reach all parts of the bunch as well as penetrate its interior (Mínguez et al. 2004; Valero et al. 2007a)
- To control berry splitting due to rain just before harvest (JECFA 2001)

### 11.2.6 OTA Control Strategies in Wine

The factors affecting the levels of OTA in wines are systematised in Fig. 11.3. The presence of OTA in wines results exclusively from grape contamination. Thus, the

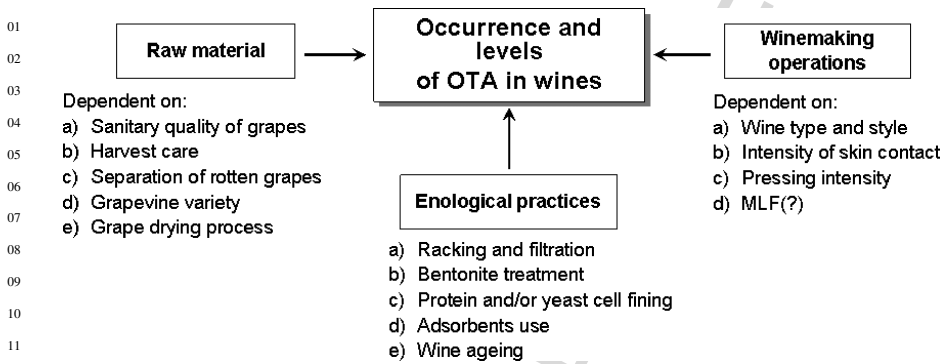


Fig. 11.3 Factors affecting the levels of OTA in wines

main measure to prevent the entry of grapes with high levels of OTA in wineries is to reject the rotten bunches and particularly those presenting brown or black rot. This is difficult to implement during harvest because rot may not be easily visible due to its development in the inner part of the bunch. In fact, bunch selection is only effective in small wineries, where the harvest is by hand and the personnel may be well trained. For large cooperatives or company wineries it is not so easy to separate the rot bunches, except if grape payment can be differentiated according to its health status. OTA determination is not an easy task at winery level (Cigič et al. 2006 and references cited therein). Rapid diagnostic tools are being developed but have not yet reached industrial laboratories (Magan 2006). This drawback can, however, be overtaken by the use of instruments that provide indicators of grape composition and health status, like the modern and costly FTIR (Fourier Transformed Infrared) spectroscopy. This method gives a measure of the grape rot that may be an indicator of the presence of OTA because gluconic acid and glycerol are correlated with *Botrytis*, sour rot and *Aspergillus* attack (Mínguez et al. 2004). When mechanical harvesting is used, it is impossible to reject rot bunches. In this case, the only measure is to control analytically the health quality of grapes in order to process the poor quality grapes separately.

Despite the fact that South African and Australian red and white wines did not show different OTA incidence (Leong et al. 2006a), most studies indicate that red wines, in the same region, contain higher OTA concentrations than rosés and white wines (for references see review of Varga and Kozakiewicz 2006). This may be explained by OTA release from grape skins (Otteneder and Majerus 2000) or by fungal spore release (Atoui et al. 2007) during red grape maceration. Therefore, wine-making processes should significantly influence its content. Several studies showed that OTA content in wines increased with the maceration time and decreased with solid-liquid separation steps, such as red wine racking or clarification of white juices (Fernandes et al. 2003; Leong et al. 2006b). The reported reductions may range from 50% to 80% of initial OTA concentrations (Hocking et al. 2007). In recent work performed with grapes from Northwest Portugal, Fernandes et al. (2007) showed

01 that, in different red vinification trials, the mean carry-over of OTA from artificially  
02 infected grapes to wine was 8.1 wt% after malolactic fermentation, even without  
03 use of enological adjuvants (fining agents), corroborating the findings of Leong  
04 et al. (2006b). Reduction of OTA was associated with removal of spent fractions  
05 during winemaking, such as wine lees after fermentation or sediment after racking,  
06 in which OTA contents were higher than in the original grapes. Some studies were  
07 performed to assess the effect of enological adjuvants, such as bentonite, gelatin,  
08 charcoal, and yeast cell wall preparations on the removal of OTA from wines (Leong  
09 et al. 2006b; Mateo et al. 2007). Most of them reduce the OTA content of wines, but  
10 the necessary concentrations have a strong effect on the wine quality. In addition, up  
11 to 29% of OTA spontaneous reduction has been observed during wine storage over  
12 10–14 months (Hocking et al. 2007).

13 Little is known about the eventual degradation or binding of OTA by yeasts or  
14 lactic acid bacteria during the fermenting process, though this has been demon-  
15 strated as possible (Angioni et al. 2007; Hocking et al. 2007). However, when  
16 compared with the physical removal of OTA during the vinification, this practice  
17 is quantitatively irrelevant. So, the most important measures for a Code of Good  
18 Enological Practices to prevent or reduce the OTA content in wines are:

- 19 – To train the harvesting staff to reject rotten bunches, particularly those affected  
20 by dark brown or black moulds
- 21 – To vinify separately the mechanical harvested grapes, when the sanitary quality  
22 of the crop is poor
- 23 – To monitor, in large wineries and cooperatives, the sanitary quality of the grapes  
24 by FTIR instruments in order to (i) favour with fair pricing the sanitary status of  
25 grapes, and (ii) process the grapes accordingly
- 26 – To avoid long periods of maceration and to use enological adsorbents, such as  
27 activated charcoal or yeast hulls, in red wine, and bentonite, in white wine, when  
28 the crop has a relevant percentage of rotten grapes
- 29 – To perform a rapid grape drying and avoid water condensation overnight for  
30 grapes used in the vinification of dessert wines
- 31 – To implement a complete HACCP plan, from the vine to the bottled wine or  
32 raisin, in the wine regions where the OTA occurrence is higher

### 36 11.3 Volatile Phenols

37  
38 Volatiles phenols (VP) are secondary metabolites produced by yeasts, moulds and  
39 bacteria which affect the flavour of several fermented food commodities (Loureiro  
40 and Malfeito-Ferreira 2006). These molecules have been under study since the  
41 first detection in fermented grains (Steinke and Paulson, 1964). Later, Dubois and  
42 Brulé (1970) reported their presence in wines and presently the importance of VP is  
43 mainly due to their role in the mediatic “horse sweat” taint in red wines. Available  
44 toxicological data suggest that VPs do not warrant concerns about acute or long-  
45 term effects (Rayne and Eggers 2007).

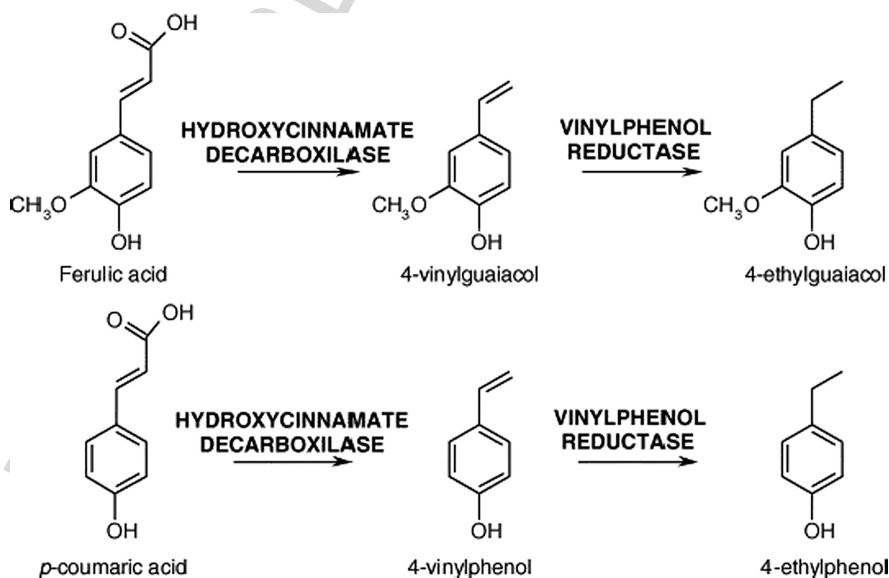
### 11.3.1 Chemical Structure and Occurrence

01 Volatile phenols are a group of molecules included in the non-anthocyanin phenolic  
 02 compounds of white and red wines. The chemical structure of VP are charac-  
 03 terised by a phenolic ring and a radical with different compositions (Fig. 11.4).  
 04 The main volatile phenols of wines are 4-vinylphenol (4-VP), 4-vinylguaiacol  
 05 (4-VG), 4-vinylcatechol (4-EC), 4-vinylsyringol (4-VS) and their respective reduced  
 06 forms 4-ethylphenol (4-EP), 4-ethylguaiacol (4EG), 4-ethylcatechol (4-EC) and  
 07 4-ethylsyringol (4-ES). These molecules occur in wines with different concentra-  
 08 tions, being vinylphenols typically associated with white wines and ethylphenols  
 09 with red wines. Table 11.3 shows recent results on the average concentration range  
 10 of these molecules in wines. In addition, vinylphenols may be also present in red  
 11 wines as anthocyanin-vinylphenol adducts, like malvidin 3-glucoside-4-vinylphenol  
 12 (Cameira-dos-Santos et al. 1996) and similar 4-VP, 4-VC and 4-VS derivatives  
 13 (Schwarz et al. 2003; Suárez et al. 2007).

### 11.3.2 Origin

#### 11.3.2.1 Availability of Precursors

22 The precursors of VPs are hydroxycinnamic acids which are enzymatically decar-  
 23 boxylated by a cinnamate decarboxylase, leading to vinylphenol derivatives, and

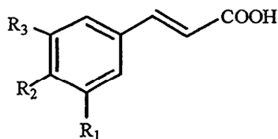


44 **Fig. 11.4** Conversion of hydroxycinnamic acids to vinylphenols and ethylphenols (reprinted from  
 45 Suárez et al. 2007, permission to be obtained)

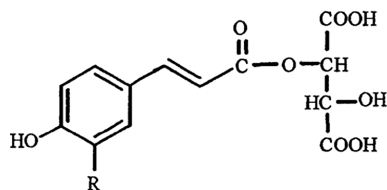
Table 11.3 Range of volatile phenol concentrations in wines ( $\mu\text{g/L}$ )

Wine type	Volatile phenols				Observations	Reference
	4-Vinylguaiacol	4-Vinylphenol	4-Ethylguaiacol	4-Ethylphenol		
White	200–324	1341–2802	Nd <sup>a</sup> -238	Nd-228	5 samples	Mejías et al. (2003)
Red	Nd-880	Nd+2174	72–255	97–782	5 samples	López et al. (2002)
	5.4–236	8.1–98	0.53–420	8.6–1500	57 aged red wines	
	–	–	46.6–169.3	399.5–2231.6	18 red wines	Tat et al. (2007)
	–	–	4.3–410.5	<0.5–586.2	54 red barrel aged wines	Rayne and Eggers (2007)
	–	–	Nd-60	127–494	9 tainted Pinot Noir wines	Hesford and Schneider (2004)
Sherry	–	–	Nd-329	298–3780	11 tainted red wines	Domínguez et al. (2002)
	820–1170	800–2000	1000–2000	300–550	4 wines	

<sup>a</sup> Not detected

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Hydroxycinnamic acid	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>
<i>p</i> -Coumaric	H	OH	H
Caffeic	OH	OH	H
Ferulic	OCH <sub>3</sub>	OH	H
Sinapic	OCH <sub>3</sub>	OH	OCH <sub>3</sub>



Hydroxycinnamic ester	R
<i>Trans</i> -caffeyltartaric acid (caftaric acid)	OH
<i>Trans-p</i> -coumaroyltartaric acid (cutaric acid)	H
<i>Trans</i> -feruloyltartaric acid (fartaric acid)	OCH <sub>3</sub>

16 **Fig. 11.5** Hydroxycinnamic acids and their esters (obtained from Monagas et al. 2005, permission  
17 to be obtained)

18  
19  
20 reduced by a vinylphenol reductase, originating the ethylphenol derivatives (Heresz-  
21 tyn 1986a and Fig. 11.4). In grape juices, hydroxycinnamic acids are esterified,  
22 mainly to tartaric acid (Fig. 11.5). In wines they may be present in the free or ester-  
23 ified form, either with tartaric acids or other polyphenols (Table 11.4). Recent  
24 work has shown that most of the free acids from tartaric esters appear after the malolactic  
25 fermentation (Hernández et al. 2006, 2007; Cabrita et al. 2007). The other esterified  
26 forms of hydroxycinnamic acids are cinnamoyl-glucoside anthocyanins (Romero  
27 and Bakker 2000; Monagas et al. 2005; Oliveira et al. 2007) and *trans-p*-coumaric  
28 acid hexoses (Monagas et al. 2005; Hernández et al. 2006, 2007). Boselli (2006)  
29 also mentioned the existence of hydroxycinnamic acid esters with ethanol, like ethyl  
30 caffeoate, in white wines.

31 The above-mentioned release of hydroxycinnamic acids from anthocyanin esters  
32 during wine maturation may be only due to chemical reactions but conversions of  
33 acid precursors to volatile phenols are typically dependent on enzyme or microbial  
34 activity.

35 In grapes or grape juices, the tartaric esters may be hydrolysed by enzymes  
36 from contaminant fungi or from commercial pectolytic preparations, both with cin-  
37 namoyl decarboxilase activity, releasing free hydroxycinnamic acid forms (Dugelay  
38 et al. 1993; Gerbaux et al. 2002). However, the tartaric esters are mostly hydrolysed  
39 after malolactic fermentation (Hernández et al. 2006, 2007), it being hypothesised  
40 that the hydrolytic activity of lactic acid bacteria follows the completion of malic  
41 conversion to lactic acid (Cabrita et al. 2007) (see Table 11.4).

42 It is accepted that yeasts only metabolise the free acid forms, although brewing  
43 *S. cerevisiae* was supposed to possess feruloyl esterase activity (Coghe et al. 2004).  
44 Then the availability of free hydroxycinnamic acids appears to be crucial for the  
45 production of VPs either by yeasts or bacteria.

Table 11.4 Range of concentrations of hydroxycinnamic acids and their esters (mg/L)

Hydroxycinnamic acids and their esters		Hydroxycinnamic acids and their esters		Observations	Reference
<i>trans</i> -Cafataric	<i>trans</i> -Caffeic	<i>trans</i> -Coutaric	2-Coutaric	Malvidin-(6-coumaroyl)-3-glucoside	Iberm-Gómez et al. (2002)
12.9–34.3	5.5–11.8	6.1–22.1	3.1–6.6	6.7–43.1	
Cafataric	Caffeic	Coutaric	Fertaric	Syringic	Pérez-Margariño and González-San José (2005)
8.89–26.06	2.36–9.95	5.35–10.04	0.40–2.14	7.08–10.61	
Cafataric	Caffeic acid	Coutaric	Coumaric acid	6 red wines not aged	Makris et al. (2006)
4.3–122.3	0–31.2	2.8–39.0	0–7.1	41 young red wines	
Cafataric	Caffeic	<i>p</i> -Coutaric	<i>m</i> -Coutaric	Ferulic	Gómez-Míguez et al. (2007)
13.43–25.29	0.04–0.98	0.76–3.59	0.12–0.14	0.13–0.04	
27.13–57.14	0.78–4.11	3.08–4.94	1.03–0.48	0.58–1.44	
<i>trans</i> -Cafataric	<i>trans</i> -Caffeic	<i>trans</i> -Coutaric	<i>trans</i> - <i>p</i> -Coutaric	<i>trans</i> - <i>p</i> -coumaric hexose	Gómez-Míguez et al. (2007)
14.98	2.68	13.75	Coumaric	1.23	
6.85	26.24	8.68	2.06	1.84	
<i>cis</i> -Cafataric		<i>cis</i> -Coutaric	16.01	Red wine before MLF <sup>a</sup>	
2.11			<i>cis</i> - <i>p</i> -Coutaric	Red wine after MLF	
0.18		4.33	1.91	Red wine before MLF	
Cafataric	Caffeic	2.17	0.52	Red wine after MLF	Cabrera et al. (2007)
4.10	0.47	Coutaric	<i>p</i> -Coumaric		
0.25	6.90	0.85	0.24	Ferulic	
		0.12	2.31	–	
			0.06	0.95	

<sup>a</sup>Malolactic fermentation



### 11.3.2.2 Conversion of Hydroxycinnamic Acids

Once hydroxycinnamic acids are released in the grape juice or in the wine, they may be converted into vinylphenol and ethylphenol derivatives depending on the presence of specific growing microbial populations.

During fermentation, *S. cerevisiae* may produce vinylphenol derivatives due to the presence of cinnamate decarboxylase enzymes (Chatonnet et al. 1992, 1993) which are inactive in red juices due to the polyphenol components of red wine (Chatonnet et al. 1997). Several grape juice contamination yeast species also have the ability to form vinylphenols (Dias et al. 2003a) but their contribution to the vinylphenol content of wines may only be relevant when are not inhibited by *S. cerevisiae* (Barata et al. 2006).

After the decarboxylation step, vinylphenols may be reduced to ethylphenols but the sequential decarboxylase and reductase activities, regarding wine yeasts, have only been demonstrated in *D. bruxellensis* and in *P. guilliermondii* (Barata et al. 2006). The former species may also convert 4-VP into 4-EP in the absence of hydroxycinnamic acids (Dias et al. 2003b).

The production of 4-ethylphenol in wines is dependent on the presence of growing yeast populations. As *P. guilliermondii* does not grow in wines with average ethanol of 12 vol.% it is not likely to produce significant levels of 4-EP during wine maturation. Concerning *D. bruxellensis* it does not grow in white wines explaining the absence of phenolic taint in this type of wines (Malfeito-Ferreira et al. 2001). Several lactic acid bacteria (*Lactobacillus* spp., *Pediococcus* spp.) have also been characterised concerning the production of ethylphenols in synthetic media (Table 11.5) but in wines they are not regarded as significant 4-EP producers (Chatonnet et al. 1995, 1997). The main starter used in wines for malolactic conversion, *Oenococcus oeni*, does not seem to produce vinyl or ethylphenols even in synthetic media (Couto et al. 2006).

The conversion of *p*-coumaric acid into 4-EP only occurs when *D. bruxellensis* is growing on a carbon and energy source, the conversion rate being dependent on the substrate (Dias et al. 2003b). The conversion of the other hydroxycinnamic acids by yeasts has not been deeply studied. Most studies are related with *p*-coumaric acid metabolism but the conversion of ferulic, caffeic acids or sinapic acids may not be equally efficient, as demonstrated, in synthetic medium, for *D. bruxellensis* (Heresztyn 1986a), *S. cerevisiae* (Chatonnet et al. 1989) and *D. anomala* (Edlin et al. 1995). Knowing that caffeic acid is more concentrated than *p*-coumaric acid in wines (see Table 11.5) it would be expected that 4-EC would be present in higher concentration than 4-EP, but the few results published do not corroborate this hypothesis (see Table 11.4). Then, *D. bruxellensis*, although utilising caffeic acid (Heresztyn 1986), may not produce 4-EC with the same efficiency as 4-EP in wines.

AQ6

### 11.3.2.3 Changes in Wine Composition

The above-mentioned metabolic activities of microorganisms should be taken into account when studying wine compositional alterations. In fact, the effect of

**Table 11.5** Metabolic activity of microorganisms related with production of volatile phenols in wine industry

Species	Function	Metabolic activity	Reference
<i>Aspergillus niger</i>	Mould for commercial enzyme production (pectinase, hemicellulase)	Active cinnamoyl esterase releasing free hydroxycinnamic acids in juices	Dugelay et al. (1993)
<i>Saccharomyces cerevisiae</i>	Fermenting yeast	Active hydroxycinnamate decarboxylase producing vinylphenols in fermenting white juices	Dugelay et al. (1993); Chatonnet et al. (1989); Shinohara et al. (2000); Barata et al. (2006)
<i>Dekkera bruxellensis</i>	Spoilage yeast	Active hydroxycinnamate decarboxylase and vinylphenol reductase producing ethylphenols in synthetic media, juices and wines	Heresztyń (1986); Chatonnet et al. (1995); Shinohara et al. (2000); Rodrigues et al. (2001); Dias et al. (2003a, 2003b)
<i>Pichia guilliermondii</i>	Contamination yeast	Active hydroxycinnamate decarboxylase and vinylphenol reductase producing ethylphenols in synthetic media and grape juices	Barata et al. (2006)
<i>C. albidos</i> , <i>C. laurentii</i> , <i>C. stellata</i> , <i>C. wickerhamii</i> , <i>D. hanseni</i> , <i>H. anomala</i> , <i>H. ivarum</i> , <i>K. apiculata</i> , <i>K. thermotolerans</i> , <i>M. pulcherrima</i> , <i>P. guilliermondii</i> , <i>P. membranifaciens</i> , <i>R. rubra</i> , <i>S. pombe</i> , <i>Z. bailii</i>	Contamination yeasts	Active hydroxycinnamate decarboxylase activity producing vinylphenols in synthetic media and grape juices	Chatonnet et al. (1992); Shinohara et al. (2000); Dias et al. (2003a)
<i>Lactobacillus</i> spp., <i>Pediococcus</i> spp.	Fermenting and spoilage lactic acid bacteria	Active hydroxycinnamate decarboxylase and vinylphenol reductase producing ethylphenols in synthetic media	Cavin et al. (1993); Couto et al. (2006)
<i>Oenococcus oeni</i> , <i>Lactobacillus plantarium</i>	Fermenting lactic acid bacteria	Active cinnamoyl esterase releasing free hydroxycinnamic acids in red wines	Hernández et al. (2007); Cabrita et al. (2007)

01 microorganisms on the polyphenolic composition does not seem to have been con-  
02 sidered when establishing differentiation between grape varieties (Makris et al. 2006),  
03 between ripening stages (Pérez-Magariño and González-San José 2005) or bot-  
04 tle aging (Monagas et al. 2005). However, as observed in Table 11.4, changes in  
05 hydroxycinnamic acid compositions may be explained by microbial activity, which  
06 are higher than changes between grape varieties or ripening stages. These variations  
07 may explain, at least partially, the controversy on the evolution of hydroxycinna-  
08 mates during wine processing (Monagas et al. 2005 and references cited therein).

09 The balance of VP and precursors is also influenced by non-microbial reac-  
10 tions. The esterified forms of hydroxycinnamic acids or vinylphenols form a pool  
11 of molecules which release or combine the acids during wine maturation, appar-  
12 ently without the influence of microorganisms (Hernández et al. 2006, 2007; Suárez  
13 et al. 2007). In addition, hydroxycinnamic acids may suffer oxidative condensation  
14 and browning during aging (Yokotsuka and Singleton 2001). Oak chips may also  
15 release 4-EG up to 0.15  $\mu\text{g/g}$ , or 4-VG up to 7.76  $\mu\text{g/g}$ , as influenced by higher  
16 toasting intensity (Natali et al. 2006). Overall, sources of VPs other than microbial  
17 should not account by more than 100  $\mu\text{g/L}$  (Rayne and Eggers 2007).

AQ7

### 11.3.3 Effect of Volatile Phenols on Product Quality

23 The sensorial effect of a volatile compound may be positive or negative to wine  
24 depending on its smell and concentration. In wines it is not easy to define beneficial  
25 or detrimental effect because the odours of mixtures of different compounds are  
26 perceived differently than those of single compounds and there is also a matrix  
27 effect on the perception. In addition, the rejection of an odour occurs at higher  
28 concentrations than the detection, leading to different detection and preference  
29 thresholds. The value of a detection/preference threshold may measure the spoilage  
30 effect of molecules with sensorial activity. These may be defined as the minimum  
31 concentration under which 50% of the tasters, in a 70 person jury, statistically  
32 detected/rejected the sample (Chatonnet et al. 1992). For instance, in Bordeaux red  
33 wines, the preference threshold for 4-EP is about 620  $\mu\text{g/L}$ , and for the mixture  
34 (10:1) of 4-EP and 4-EG is 426  $\mu\text{g/L}$  (Chatonnet et al. 1992). Below these concen-  
35 trations, volatile phenols may contribute favourably to the complexity of wine aroma  
36 by imparting aromatic notes of spices, leather, smoke or game, appreciated by most  
37 consumers. Above those levels, wines are clearly substandard for some consumers  
38 but remain pleasant for others. To increase the difficulty in the definition of spoiling  
39 concentrations, these thresholds are dependent on grapevine variety and on the style  
40 of wine (Gato et al. 2001; Coulter et al. 2003).

41 In the case of vinylphenols, they contribute to the spicy, floral and pharmaceutic  
42 character of white wines. The 4-VG has been detected in high levels in the variety  
43 Gewürztraminer (Grando et al. 1993). The depreciation due to high levels of 4-VG  
44 plus 4-VP in white wines of the German variety Kerner was associated with hot  
45 regions (e.g. South Africa) or exposure of grapes to sunlight, but no explanation

01 was given to the fact (Rapp 1998). The overall incidence of vinylphenols in white  
02 wines is not known and but it seems to have decreased after the improvement in the  
03 purity of commercial pectolytic enzymes used in juice clarification.

04 In the case of ethylphenols, concentrations of 4-EP and 4-EG above the prefer-  
05 ence thresholds dominate the flavour contributing to the phenolic character. The  
06 mouthfeel sensations are also altered by increasing the metallic notes (Coulter  
07 et al. 2003). Volatile phenols are currently determined by gas-liquid chromatography  
08 after wine extraction with organic solvents (Loureiro and Malfeito-Ferreira 2006).  
09 However, the main flaw of this technique is the absence of 4-EC quantification,  
10 which requires derivatisation (Hesford and Schneider 2004). Despite this fact,  
11 numerous recent improvements in volatile phenol analysis were only directed to  
12 extraction procedures (López et al. 2002; Mejías et al. 2003; Díez et al. 2004; Fariña  
13 et al. 2007; Pizarro et al. 2007; Rayne and Eggers 2007). In contrast, Carrillo and  
14 Tena (2007) presented an HS-SPME extraction followed by GC-MS of derivatised  
15 samples accounting for 4-EC. In addition, liquid chromatography has been present  
16 as an alternative to GC, having the advantage of avoiding sample extraction (Van-  
17 beneden et al. 2006; Caboni et al. 2007; Larcher et al. 2007; Nicolini et al. 2007)  
18 but the proposed methods do not account for 4-EC.

19 The real incidence of VP in wines world wide is not known perhaps due to  
20 the difficulty in performing routine instrumental analysis in wineries. Normally,  
21 winemakers analyse samples suspected to have problems and so the reported pro-  
22 portion of affected wines is most probably biased. Some data from analytical  
23 laboratories have shown that from 6% to 74% of analysed samples may bear levels  
24 of 4-EP plus 4-EG higher than the preference threshold (426 µg/L) of a 10:1 mixture  
25 (Loureiro and Malfeito-Ferreira 2006). Although precise numbers are not available,  
26 we believe that the phenolic taint is the main microbiological problem leading to  
27 higher economical losses in winemaking industry. Moreover, as already mentioned,  
28 the sensory detection of VP depends on the type of wine (Gato et al. 2001; Coulter  
29 et al. 2003) and so a higher proportion of wines may be badly affected by these  
30 compounds.

31 The 4-EG is present in about one tenth of the 4-EP concentration (Chatonnet  
32 et al. 1992) but this rate is not always observed (Rodrigues et al. 2001; Coulter  
33 et al. 2003). The other odour active VP, 4-EC, has just begun to be studied. The  
34 fact that the precursors of 4-EC, caffeic acid and its esters, are present in rela-  
35 tively high concentrations in wines (see Table 11.4) and that the detection threshold  
36 for 4-EC (described as having a phenolic smell similar to that of 4-EP) is about  
37 50 µg/L (Porret et al. 2004) suggests that its influence in phenolic taint should be  
38 not be neglected. Occasional discrepancies between sensorial detection and con-  
39 centration of 4-EP and 4-EG may be explained by the hidden presence of 4-EC.  
40 Sinapic acid gives 4-VS and 4-ES but the syringols do not affect medium odour  
41 (Heresztyn 1986a).

42 The wines affected by ethylphenols are practically only red wines. All types are  
43 susceptible to the phenolic taint depending on the growth of *D. bruxellensis*. How-  
44 ever, red wines matured in oak barriques are the typical wine product affected by  
45

01 this taint because these containers provide a highly favourable ecological niche for  
02 *D. bruxellensis* (Loureiro and Malfeito-Ferreira 2006).

### 05 **11.3.4 Control Measures**

07 When wines are affected by volatile phenols there is, at present, no effective cura-  
08 tive process. In this situation, oenologists always weigh the possibility of blending  
09 tainted wine with “clean” wine. Although this measure may attenuate the defect  
10 of the tainted wine by dilution it cannot be seen as a curative measure. In fact,  
11 mixtures of wines with null or low levels of 4-ethylphenol are only effective for  
12 small proportions of tainted wines because large volumes of “clean” wine must  
13 be used to obtain a blend with 4-EP levels lower than the preference thresh-  
14 old. Then, effective curative measures would depend either (i) on the reduction  
15 or elimination of the sensorial effect or (ii) on the extraction of odour active  
16 molecules from the wine. These strategies have not yet been effectively tested in  
17 practical conditions. Guilloux-Benatier et al. (2001) hypothesised that yeast lees  
18 have the property to adsorb volatile phenol, which was later shown, at least par-  
19 tially, in laboratory conditions, by Chassagne et al. (2005). A reverse osmosis  
20 procedure to reduce volatile phenols is commercially available in New Zealand  
21 (<http://www.armourtech.co.nz/memstarreverseos.html>). Salameh et al. (2007)  
22 showed that *p*-coumaric acid may be adsorbed on *Brettanomyces* cells, decreas-  
23 ing its availability as substrate. Commonly, when adsorbents are added to wine,  
24 favourable aroma compounds are also removed and a balance must be drawn  
25 between benefits and losses of wine attributes.

26 If a curative approach is not effective, then prevention is, at present, the most  
27 reasonable way to deal with the problem. Bearing in mind that, to produce volatile  
28 phenols microorganisms needs the substrate to be available and active, the preven-  
29 tive measures may be directed either to minimise the release of free acid or to avoid  
30 microbial activity.

31 In juices, prevention should be based on (i) decreasing the release of free acids  
32 that is favoured by mould infections of grapes and by the decarboxylase activity of  
33 commercial enzyme preparations and (ii) avoiding the production of volatile phenols  
34 that is favoured by the uncontrolled activity of contamination yeasts growing in  
35 damaged grapes or in juices. Then, the main measures to be adopted are:

- 37 – Separation of sound grapes from damaged grapes
- 38 – Use of sulphur dioxide to prevent yeast contaminations
- 39 – Use of pure commercial enzymes, if necessary
- 40 – Initiation of active fermentation with *S. cerevisiae* as soon as possible

42  
43 These measures are good manufacturing practices of winemaking, irrespective of  
44 the risk of phenolic taint. The main preventive measures should be performed during  
45 wine storage, aging (mainly in oak barrels) and bottling. As the release of precursors

01 is unavoidable, the main preventive measures are directed to reduce the activity of  
02 contaminating populations of yeasts and bacteria, especially towards *D. bruxellensis*  
03 (see review of Loureiro and Malfeito-Ferreira 2006):

- 04
- 05 – Use adequate hygienic practices and respective efficiency assessment
- 06 – Use adequate levels of sulphur dioxide or DMDC (if legally authorised)
- 07 – Minimise residual nutrient contents (sugar or nitrogen)
- 08 – Minimise oxygen dissolution
- 09 – Handle oak barrel aging properly (disinfection, toppings, rackings, cellar temper-  
10 ature)
- 11 – Perform microbiological monitoring especially when wines from external sources  
12 are processed and oak aging is used
- 13 – Thermal treatment or filter sterilisation is advised when risk of bottle infection is  
14 high
- 15
- 16

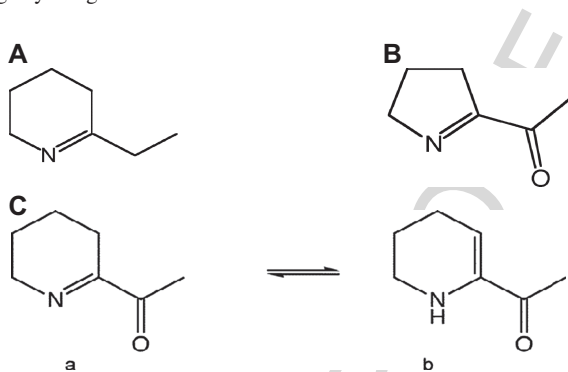
## 17 **11.4 Tetrahydropyridines**

18  
19 Tetrahydropyridines (THP) are secondary metabolites produced by *D. bruxellensis*  
20 and lactic acid bacteria in wines and are responsible for a taint described as mousy  
21 off-flavour or mousiness. This problem has been known since late nineteenth cen-  
22 tury (see review of Snowdon et al. 2006) but, in spite of its obnoxious flavours, has  
23 been only vaguely studied perhaps due to its low frequency of occurrence.

### 24 **11.4.1 Chemical Structure and Origin**

25  
26  
27  
28  
29 Tetrahydropyridines (THP) include 2-ethyl-tetrahydropyridine (ETHP), 2-acetyl-  
30 tetrahydropyridine (ATHP) and 2-acetylpyrroline (APY) (Fig. 11.6). ETHP is present  
31 in tautomeric forms, but the second tautomer is minor. ATHP also occurs in two  
32 tautomeric forms, of which the distribution is pH dependent. These molecules are  
33 uncommon components of wines and are not currently analysed.

34 Tucknott (1977) first identified ETHP and other unknown compounds as the  
35 molecules imparting mousiness, showing that they were not present in the absence  
36 of microorganisms. The origin of THP is related with activity of *D. bruxellensis*  
37 and of lactic bacteria, mainly heterofermentative strains, but the possible role of  
38 acetic bacteria should not be discarded (Heresztyn 1986b; Snowdon et al. 2006).  
39 APY is not produced by *Dekkera* spp. but by lactic acid bacteria, being an indica-  
40 tor of bacterial spoilage. The pathways for production of THP by heterofermenta-  
41 tive bacteria or *D. bruxellensis* have been proposed (Costello and Henschke 2002;  
42 Snowdon et al. 2006) but require further confirmation. However, it is established  
43 that both pathways require L-lysine and ethanol to THP synthesis. L-Lysine and  
44 L-ornithine are responsible for ring formation of the heterocycles whereas ethanol  
45 and acetaldehyde are responsible for the acetyl side chain (Snowdon et al. 2006;



**Fig. 11.6** Chemical structure of (A) 2-ethyltetrahydropyridine (ETHP), (B) 2-acetylpyrroline (APY) and tautomers of (Ca) 2-acetyl-3,4,5-tetrahydropyridine (ATHP) and (C) 2-acetyl-1,4,5,6-tetrahydropyridine (adapted from Snowdon et al. 2006, permission to be obtained)

Grbin et al. 2007). ATHP reduction may lead to EHTP. As ethanol is a precursor, mousy off-flavour occurs after alcoholic fermentation, preferably after lactic acid bacteria activity. It seems that the formation of mousiness may be induced by oxidation but it is not clear if the effect is on the microorganisms or in any chemical reaction stimulated by the redox potential. Other agents claimed to affect its production (high pH, low sulphite, residual sugar content) (Lay 2004; Snowdon et al. 2006; Romano et al. 2007) are also stimulators of microbial activity and so the true mechanisms are not yet clarified, but the non-enzymatic chemical synthesis has been ruled out in *D. anomala* (Grbin et al. 2007).

#### 11.4.2 Effect on Wine Quality and Occurrence

The “mousy off-flavour” is described as resembling the smell of mice urine and once tasted becomes unforgettable. The taint is mainly perceived by the after-mouth sensation and has a long persistence (may exceed 10 min). The compounds responsible are not volatile at wine pH and so are only perceived by the increase in pH due to saliva. These features justify the use of practical sensorial detection methods without the need to swallow the wine, like rubbing the wine in the palm of the hand and sniffing the skin, or dipping an alkaline paper strip in the wine and smelling. ATHP is the main molecule responsible for the fault (Strauss and Heresztyn 1984), being present in levels up to 108  $\mu\text{g/L}$  (Snowdon et al. 2006) and having a detection threshold of 1.6  $\mu\text{g/L}$  in water (Colagrande et al. 1988). The imino tautomeric form provides the mousy perception and its prevalence at high pH explains the detection after increasing sample pH. Its off-flavour also resembles cracker biscuit, and this molecule is also present in some cereal based products (Snowdon et al. 2006). EHTP has been detected in wines only recently and has a detection threshold of 150  $\mu\text{g/L}$  in wines (Snowdon et al. 2006). APY is also a major contributor to mousy off-flavour, with detection threshold of 0.1  $\mu\text{g/L}$  in water and being detected in

01 trace levels up to 7.8  $\mu\text{g/L}$  (Snowdon et al. 2006). In other food products it has been  
02 described as “roasty” and “popcorn-like”. Acetamide, although occasionally linked  
03 to mousiness, is not the cause of mousy off-flavour because it is odourless (Snowdon  
04 et al. 2006).

05 The incidence of mousiness in wines is not known. In our experience we have  
06 only tasted it a handful of times. Earlier reports and classical enology treatises  
07 already mention this problem (Grbin and Henschke 2000 and references cited  
08 therein). The activity of *Dekkera/Brettanomyces* spp. has been linked to this fault  
09 since early studies leading to some confusion between mousiness and phenolic taint.  
10 The development of each other is independent (Romano et al. 2007) and we are  
11 not aware of the factors stimulating one instead of the other. Although it is not a  
12 common taint, it is known to affect wines all over the wine countries (Grbin and  
13 Henschke 2000). As it depends on *Dekkera/Brettanomyces* activity it is likely that  
14 red wines are more affected. Also, white wines are usually not subjected to malo-  
15 lactic fermentation and so the activity of lactic bacteria may preferentially affect red  
16 wines. These yeasts also produce THP in grape juices (Grbin and Henschke 2000)  
17 but the real incidence in this product is not known.

### 20 **11.4.3 Control Measures**

22 There is no available method to remove this taint effectively (Lay 2004). The  
23 removal of precursors (L-lysine and ethanol) is not feasible. As it depends on micro-  
24 bial activity, the preventive measures are similar to those suggested for volatile phen-  
25 nols when there is the risk of *D. bruxellensis* infection. The prevention of spoilage by  
26 heterofermentative lactic bacteria usually advised, like decreasing wine pH values  
27 and rapid inactivation by sulphur dioxide, once malolactic conversion is finished,  
28 should also be effective against bacterial mousiness.

## 32 **11.5 Final Remarks**

34 The wine spoilage effects described in this chapter illustrate two different approaches  
35 undertaken to solve the problems. One, OTA production, is associated with food  
36 safety and so profits from the allocation of relatively large funds under specific  
37 EU research frameworks. The other problems, associated mainly with VP, are of  
38 technological nature and are not specifically supported by research frameworks and  
39 so the respective scientific outputs are much less relevant. The present awareness of  
40 OTA incidence and knowledge of preventive measures make this a relatively minor  
41 problem to the wine industry, which can easily be kept under control by using an  
42 adequate HACCP system from vineyard to the bottle. In contrast, concerning phe-  
43 nolic taint, this continues to be a major problem of wine microbial spoilage which  
44 involves significant economical losses. The primary sources and routes of contam-  
45 ination of *D. bruxellensis*, the role of esterified VP precursors, the role of lactic



acid bacteria, the white wine enigmatic resistance to VP production, the reasons explaining unexpected blooms, the choice of effective prevention measures, and the different wine susceptibility to spoilage are only some of the issues not fully understood by scientists and wine technologists. From a practical point of view, the slow response of the production sector to the problem is surprising, particularly concerning *D. bruxellensis* monitoring in wines and equipment surfaces. The utilisation of microbiological control and the adoption of guidelines by attributes for bulk and bottled wines would certainly avoid most problems, given that the essential of the knowledge on *D. bruxellensis* spoiling activity is already available.

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01 **Chapter-11**

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03 **Query No. Page No. Line No. Query**

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04

05 AQ1 615 07 Should this not be “implying that any pathogenic bac-

06 teria DID NOT HAVE the capacity to grow or sur-

07 vive in wine”???. Please check sense of statement as

08 it is!!!

09 AQ2 620 17 I do not understand what you mean by ‘busted’ –

10 have substituted ‘greatly increased’. Please check

11 and change as required.

12 AQ3 622 05 “Bellí et al., 2005” is not listed in the reference list.

13 Please provide.

14 AQ4 626 45 a or b? There are two such references reference list.

15 AQ5 627 TABLE 11.3 a or b? There are two such references reference list

16 AQ6 631 38 “Heresztyn, 1986” whether it is “1986a” or “1986b”?

17 AQ7 633 17 a or b? There are two such references reference list.

18 AQ8 634 13 a or b? There are two such references reference list.

19 AQ9 642 38 Please update.

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